## **Isonucleosides with Exocyclic Methylene Groups**

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Dedicated to Prof. Dr. Frank Seela on the occasion of his 60th birthday

Synthesis of isonucleosides 13, 14, 16, and 17, bearing an exocyclic methylidene group at the sugar moiety, starting from a 3-keto sugar is described. The keto compound was converted to the methylene-sugar 10b (*Scheme 1*), which was coupled with nucleobases by means of the *Mitsunobu* reaction. The coupling reaction with adenine and 8-azaadenine produced both the  $N^9$ - and  $N^3$ -nucleosides (see 13 and 14, resp.; *Scheme 2*). The structures of 13a and 14a were confirmed by single-crystal X-ray data. Synthesis of the pyrimidine compounds was also approached from the  $\beta$ -amino sugar 20 that was prepared using a *Gabriel*-synthesis methodology (*Scheme 3*).

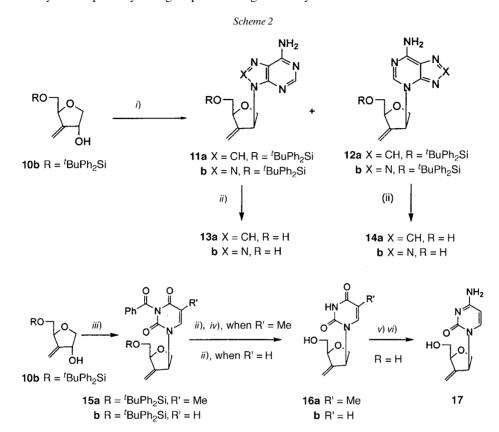
**Introduction.** – For a number of years, we have been investigating the synthesis, enzymology, and antiviral studies of various isomeric nucleosides in our laboratory [1]. For example, (2S,4S)-4-(6-amino-9H-purin-9-yl)tetrahydrofuran-2-methanol ((S,S)-isoddA; 1), an isomeric dideoxynucleoside synthesized by us, exhibits activity against HIV-1 and HIV-2 [2]. Our interest in synthesizing analogs of (S,S)-isoddA (1) led us to the design of isomeric dideoxynucleosides 3 with an exocyclic methylene group. This synthetic design was supported by the observation of the *anti*-HIV and *anti*-HBV activity of carbocyclic nucleoside 4 [3] and the *anti*-HIV activity of the dioxolanyl compounds 5 and 6 [4]. It is likely that the antiviral activity of 4 is associated with its ability to function as a structural analog of 2'-deoxyguanosine. The exocyclic C=C bond appears to act as a functional replacement for the O-atom at that position. If this is so then our target molecules 3 (and mirror images) are related to both 4, in which the endocyclic O-atom has replaced the exocyclic OH, and to compounds 5 and 6. This paper reports the synthesis and antiviral studies of novel isonucleosides with exocyclic methylene groups.

Results and Discussion. – Synthesis of the methylene-isonucleosides was approached from the corresponding carbohydrate precursor bearing the exocyclic methylene group. The synthesis commenced with the protected D-xylose **7** [5]. Oxidation of the 3-OH group with DMSO and  $P_2O_5$  [6] and subsequent *Wittig* reaction [7][8] on the resulting ketone with NaH and (Ph<sub>3</sub>PMe)Br in DMSO produced the alkene **8**. Acid-catalyzed methanolysis [9] of the acetonide group afforded the  $\alpha$ - and  $\beta$ -D-glycosides **9**. Reductive demethoxylation was carried out by refluxing **9** with hexamethyldisilazane (HMDS) followed by treatment with Et<sub>3</sub>SiH and Me<sub>3</sub>SiOSO<sub>2</sub>CF<sub>3</sub> [9–11] in MeCN at room temperature to give **10a**. The primary OH group of **10a** was selectively protected [12] with the (*tert*-butyl)diphenylsilyl group to yield **10b**, which was the key starting material for the synthesis of both isodideoxypurine and pyrimidine nucleosides (*Scheme 1*).

## **Enantiomeric IsoddAs**

i) DMSO, P<sub>2</sub>O<sub>5</sub>, r.t. ii) NaH, DMSO, (Ph<sub>3</sub>PMe)Br. iii) HCl, MeOH. iv) Me<sub>3</sub>SiCl, HMDS. v) Et<sub>3</sub>SiH, Me<sub>3</sub>SiOSO<sub>2</sub>CF<sub>3</sub>, MeCN, r.t. vi) 'BuPh<sub>2</sub>SiCl, pyridine, 4°.

Purine isonucleosides having adenine or 8-azaadenine bases were synthesized via the *Mitsunobu* reaction [13–15]. Interestingly, treatment of **10b** with adenine in the presence of DEAD (diethyl diazenedicarboxylate) and Ph<sub>3</sub>P in dioxane at room temperature produced the  $N^9$ -isomer **11a** and its  $N^3$  isomer **12a** (Scheme 2). Similarly, compounds 11b and 12b were obtained when 10b was treated with 8-azaadenine under Mitsunobu conditions. Desilvlation of compounds 11 and 12 with NH<sub>4</sub>F in MeOH afforded compounds 13 and 14, respectively. The structures of both the  $N^9$ - and  $N^3$ isomers of the coupling products were confirmed by their multinuclear NMR and HR-MS data and particularly by their UV spectra and single-crystal X-ray data. For example, the UV spectrum of compound 13a showed a  $\lambda_{max}$  at 261 nm ( $\epsilon$  14000), whereas compound 14a exhibited a  $\lambda_{max}$  at 275 nm ( $\varepsilon$  13300). The X-ray structure of 13a (Fig. 1) confirmed unequivocally its structure and absolute configuration and showed the glycosidic bond attachment of N(9) of the base, the presence of the exocyclic methylene group, the anti-conformation about the glycosidic bond, and the endo puckering of the sugar ring at the O-atom. The X-ray structure of compound 14a also confirmed its structure and absolute configuration but showed the attachment of the base at N(3) (Fig. 2). Interestingly, the crystal structure of compound 14a showed the presence of two independent molecules in an asymmetric unit. Each formed a dimer via two H-bonds with a symmetry equivalent molecule of the other (*i.e.*, molecule A was dimerized with molecule B, and *vice versa*). The two molecules differ in conformation mainly at the primary OH group of the sugar moiety.



i) Ph<sub>3</sub>P, DEAD, purine base, dioxane. ii) NH<sub>4</sub>F, MeOH. iii) 3-benzoylthymine or uracil, Ph<sub>3</sub>P, DEAD, THF. iv) MeONa, MeOH. v) Ac<sub>2</sub>O, pyridine. vi) <sup>i</sup>Pr<sub>3</sub>SiCl, 4-(dimethylamino)pyridine, Et<sub>3</sub>N, MeCN, then NH<sub>4</sub>OH

Isodideoxypyrimidine nucleosides having an exocyclic methylene group were also synthesized by the *Mitsunobu* reaction. Thus, compound **10b**, on treatment with 3-benzoyluracil or 3-benzoylthymine under *Mitsunobu* conditions gave the desired products **15a,b** (*Scheme 2*). The protecting groups of **15a,b** were removed by treatment with NH<sub>4</sub>F or by successive treatment with NH<sub>4</sub>F and NaOMe to produce **16a,b**. It is interesting to note that in the case of the uracil derivative **15b**, only NH<sub>4</sub>F treatment was required to completely deprotect to product **16b**. Compound **16b** was converted to the corresponding cytosine derivative **17** according to a previously reported procedure [16].

Because of the relatively low yields obtained in the *Mitsunobu* coupling procedure with pyrimidine nucleosides, we also developed an approach to these compounds from the appropriate amino sugar **20** which was prepared from **10b** (*Scheme 3*). When compound **10b** was treated with phthalimide under *Mitsunobu*-reaction conditions, no

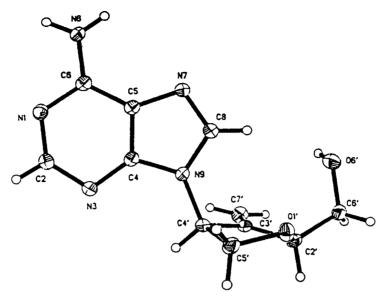


Fig. 1. ORTEP Plot of single-crystal X-ray structure of 13a

phthalimido derivative **19** was obtained. However, the reaction of mesylate **18**, prepared from **10b**, with potassium phthalimide in DMF produced stereospecifically the substituted product **19**. It should be mentioned that treatment of mesylate **18** with NaN<sub>3</sub> did not give the  $S_N2$  product but compound **22** resulting from the  $S_N2$ 

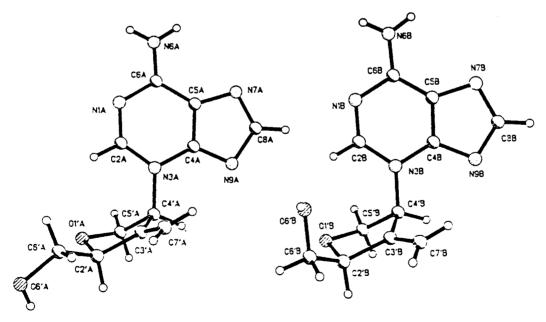


Fig. 2. ORTEP Plot of single-crystal X-ray structure of 14a

displacement reaction by the azido group. The amino derivative **20** was obtained by the deprotection of **19** with hydrazine [17]. Treatment of **20** with 3-methoxy-2-methylprop-2-enoyl isocyanate [10][18][19] produced **21**, which, on acid-catalyzed cyclization, produced the thymine derivative **16a**.

i) MsCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, 0°. ii) Potassium phthalimide, DMF, 110°. iii) NH<sub>2</sub>NH<sub>2</sub>, EtOH, reflux. iv) 3-Methoxy-2-methylprop-2-enoyl isocyanate, DMF, Et<sub>2</sub>O, 0° to r.t. v) 2N H<sub>2</sub>SO<sub>4</sub>, dioxane, 100°. vi) NaN<sub>3</sub>, DMF, 90°.

In summary, synthesis of the methylene-isonucleosides 13, 14, 16, and 17 was achieved starting from a 3-keto sugar prepared from inexpensive p-xylose. The keto sugar was converted to an appropriate methylene-bearing carbohydrate precursor that was coupled, *via* the *Mitsunobu* reaction, with nucleobases. Synthesis of the pyrimidine compound 16a was also approached from the  $\beta$ -amino sugar 20 which was prepared by means of a *Gabriel*-synthesis methodology. In the case of the direct coupling reaction with adenine and 8-azaadenine, both the  $N^9$ - and  $N^3$ -nucleosides (see 13 and 14) were produced. The structure of 13a and 14a was unequivocally confirmed by single-crystal X-ray data. Preliminary antiviral evaluations suggested that the target compounds did not show significant *anti*-HIV activity *in vitro* in infected CEM-SS cells. Further biological studies are in progress.

We are grateful to Dr. *Dale Swenson* for the single-crystal X-ray structure determinations and Dr. *Lynn Teesch* and Ms. *Diane Herschberger* for the HR-MS data. This research work was supported by the *N.I.H.* (*National Institute of Allergy and Infectious Diseases*). We thank the *National Cancer Institute* for the *anti-HIV* evaluations.

## **Experimental Part**

General. M.p.s.: uncorrected. Electrothermal-Engineering-Ltd. melting-point apparatus. UV Spectra: Cary-3 UV-Visible spectrophotometer.  $^1$ H- and  $^{13}$ C-NMR Spectra: Bruker-AC-300 instrument at 300 and 75 MHz, resp.,  $\delta$  in ppm referenced to internal SiMe<sub>4</sub> for  $^1$ H and to solvent CDCl<sub>3</sub>, (D<sub>6</sub>)DMSO, (D<sub>6</sub>)acetone, or CD<sub>3</sub>OD) for  $^{13}$ C. Column chromatography (CC): 230 – 400 mesh silica gel. High-resolution FAB-MS: VG-ZAB-HF high-resolution mass spectrometer.

3-Deoxy-1,2-O-isopropylidene-3-methylidene-5-O-trityl- $\alpha$ -D-ribofuranose (8). To a soln. of **7** [5] (4.3 g, 10 mmol) in DMSO (30 ml), P<sub>2</sub>O<sub>5</sub> (2.6 g, 18.3 mmol) was carefully added while maintaining the temp. at 20–30°. The mixture was stirred at 30° for 20 h. The brown soln. was diluted with CHCl<sub>3</sub> (120 ml) and washed with ice-cold H<sub>2</sub>O until the washings were neutral. The org. phase was dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated and the residue purified by CC: keto compound as white solid foam (2.9 g, 67%).

NaH (0.72 g, 15.1 mmol) was suspended in DMSO (30 ml) and the mixture heated at 75° for 45 min under  $N_2$ . After the soln. was cooled to r.t., methyltriphenylphosphonium bromide (5.8 g, 16.3 mmol) was added, and the soln. was stirred at r.t. for 30 min. To this mixture a soln. of the keto compound in DMSO (20 ml) was added dropwise. After the addition, the mixture was stirred for an additional 2 h and then poured into  $H_2O$  (100 ml). The aq. soln. was extracted with  $Et_2O$  (3 × 60 ml), the combined org. phase washed with  $H_2O$  (2 × 75 ml), dried ( $Na_2SO_4$ ), and evaporated, and the gummy residue purified by CC: **8** (2.3 g, 89%). White solid. M.p. 159–160°.  $^1H$ -NMR (CDCl<sub>3</sub>): 7.46–7.17 (m, 3 Ph); 5.95 (d, J = 4.1, H – C(1)); 5.34 (t, J = 1.1, 1 H, = CH<sub>2</sub>); 5.04 (t, J = 1.4, 1 H, = CH<sub>2</sub>); 4.94–4.87 (m, H – C(2), H – C(4)); 3.28 (dd, J = 9.9, 4.0,  $H_a$  – C(5)); 3.19 (dd, J = 9.9, 4.5,  $H_b$  – C(5)); 1.52 (s, 3 H,  $Me_2C$ ); 1.39 (s, 3 H,  $Me_2C$ ).  $^1S$ -NMR (CDCl<sub>3</sub>): 147.0 (C(3)); 143.7 (Ph); 128.6, 127.7, 127.0 (Ph); 112.4 ( $Me_2C$ ); 111.8 (= CH<sub>2</sub>); 104.8 (C(1)); 86.6 ( $Ph_3C$ ); 81.8 (C(2)); 79.1 (C(4)); 65.8 (C(5)); 27.4, 27.1 ( $Me_2C$ ). HR-FAB-MS: 451.1885 ( $C_{28}H_{28}NaO_4Si^+$ , [M + Na]+; calc. 451.1885).

(3R,5S)-5-[[[(tert-Butyl)diphenylsilyl]oxy]methyl]tetrahydro-4-methylidenefuran-3-ol (10b). To a soln. of 8 (5.7 g, 13.3 mmol) in MeOH (150 ml) 4n HCl in dioxane (0.27 ml) was added, and the mixture was heated under reflux for 2 h. The acid was neutralized with  $K_2CO_3$ , the suspension adsorbed on silica gel, and the mixture purified by CC: 9 (1.66 g, 78%). Oil.

A mixture of **9** (0.54 g, 3.4 mmol) and Me<sub>3</sub>SiCl (0.25 ml) in hexamethyldisilazane (HMDS; 25 ml) was heated under reflux for 17 h. Unreacted HMDS was evaporated to afford a pale yellow liquid. Dry MeCN was added to the residue followed by the addition of Et<sub>3</sub>SiH (1.6 ml, 10.2 mmol) and Me<sub>3</sub>SiOSO<sub>2</sub>CF<sub>3</sub> (1.8 ml, 10.2 mmol), and the mixture was stirred overnight at r.t. It was then quenched with H<sub>2</sub>O (5 ml), stirred for 1 h, and then neutralized with *Dowex* resin (OH<sup>-</sup> form). The soln. was filtered, the filtrate evaporated, and the residue purified by CC: (2S,4R)-tetrahydro-4-hydroxy-3-methylidenefuran-2-methanol (**10a**; 78%). Oil. <sup>1</sup>H-NMR (CD<sub>3</sub>OD): 5.29 (t, J=2.2, 1 H, =CH<sub>2</sub>); 5.11 (t, J=2.0, 1 H, =CH<sub>2</sub>); 4.55-4.48 (m, H-C(2), H-C(4)); 4.04 (dd, J=9.0, 6.1, H<sub>a</sub>-C(5)); 3.61 (dd, J=11.8, 3.5, 1 H, CH<sub>2</sub>); 3.50 (m, 2 H, H<sub>b</sub>-C(5), CH<sub>2</sub>). <sup>13</sup>C-NMR (CD<sub>3</sub>OD): 152.2 (C(3)); 108.6 (=CH<sub>2</sub>); 82.2 (C(2)); 74.2 (C(4)); 73.5 (C(5)); 65.4 (CH<sub>2</sub>).

<sup>1</sup>BuPh<sub>2</sub>SiCl (0.57 ml, 2.2 mmol) was added to a soln. of **10a** (0.26 g, 2 mmol) in pyridine (20 ml) at 0°. After the addition, the mixture was stirred at 0° for 6 h and at 4° overnight. The reaction mixture was quenched with H<sub>2</sub>O (2 ml). Pyridine was evaporated and the conc. soln. co-evaporated with toluene. The residue was purified by CC: **10b** (0.62 g, 84%). Gum. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 7.68−7.38 (m, 2 Ph); 5.34 (s, 1 H, =CH<sub>2</sub>); 5.08 (s, 1 H, =CH<sub>2</sub>); 4.60 (m, H−C(3), H−C(5)); 4.1 (dd, J =9.3, 5.6, H<sub>a</sub>−C(2)); 3.68 (m, H<sub>b</sub>−C(2), CH<sub>2</sub>); 1.04 (s, Me<sub>3</sub>C). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 151.3 (C(4)); 135.5, 135.4, 133.2, 133.1, 129.6, 129.5, 127.5 (Ph); 108.6 (=CH<sub>2</sub>); 80.6 (C(3)); 73.9, 72.9 (C(2), C(5)); 66.4 (CH<sub>2</sub>); 26.6 (Me<sub>3</sub>C); 19.1 ( $Me_3$ C). HR-FAB-MS: 391.1695 (C<sub>22</sub>H<sub>28</sub>NaO<sub>3</sub>Si<sup>+</sup>, [M + Na]<sup>+</sup>; calc. 391.1705).

(2\$,4\$)-4-(6-Amino-9H-purin-9-yl)tetrahydro-3-methylidenefuran-2-methanol (13a) and (2\\$,4\\$)-4-(6-Amino-3H-purin-3-yl)tetrahydro-3-methylidenefuran-2-methanol (14a). To a suspension of 10b (0.61 g, 1.66 mmol), Ph<sub>3</sub>P (0.65 g, 2.5 mmol), and adenine in dioxane (80 ml) at r.t., DEAD (0.4 ml, 2.6 mmol) was added dropwise. After the addition, the mixture was stirred at r.t. for 22 h. The solvent was then evaporated and the residue purified by CC: 11a (0.34 g, 42%) and 12a (0.15 g, 19%).

Data of **11a**: M.p. 170°. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 8.33 (s, H–C(8)); 7.90 (s, H–C(2)); 7.69 –7.35 (m, 2 Ph); 5.87 (s, NH<sub>2</sub>); 5.64 (br. s, H–C(4')); 5.27 (m, =CH<sub>2</sub>); 4.20 (dd, J = 10.0, 6.0, H<sub>a</sub>–C(5')); 4.14 (dd, J = 10.0, 4.0, H<sub>b</sub>–C(5')); 3.93 (d, J = 4.7, CH<sub>2</sub>). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 155.5 (C(6)); 152.9 (C(2)); 150.0 (C(4)); 146.9 (C(3')); 138.9 (C(8)); 135.6, 132.9, 129.8, 127.8 (2 Ph); 119.3 (C(5)); 112.1 (=CH<sub>2</sub>); 81.3 (C(2')); 72.1 (C(5')); 65.6 (CH<sub>2</sub>); 56.6 (C(4')); 26.8 (Me<sub>3</sub>C); 19.2 ( $Me_3$ C).

Data of 12a: M.p.  $160-161^{\circ}$ . <sup>1</sup>H-NMR (CDCl<sub>3</sub>); 8.36 (s, H-C(8)); 8.02 (s, H-C(2)); 7.70-7.26 (m, 2 Ph); 6.05 (br. s, H-C(4')); 5.50 (br. s, 1 H, =CH<sub>2</sub>); 5.40 (br. s, 1 H, =CH<sub>2</sub>); 4.50 (m, H-C(2')); 4.25 (d, J = 4.0,

 $2 \text{ H} - \text{C(5')}; 4.02 \ (dd, J = 11.5, 3.9, 1 \text{ H}, \text{ CH}_2); 3.97 \ (dd, J = 11.4, 4.4, 1 \text{ H}, \text{ CH}_2); 1.07 \ (s, \text{Me}_3\text{C}). \ ^{13}\text{C} - \text{NMR} \ (\text{CDCl}_3); 154.4 \ (\text{C(6)}); 153.7 \ (\text{C(2)}); 150.6 \ (\text{C(4)}); 145.8 \ (\text{C(3')}); 140.8 \ (\text{C(8)}); 135.5, 132.8, 129.9, 127.8 \ (\text{Ph}); 120.5 \ (\text{C(5)}); 113.9 \ (=\text{CH}_2); 81.5 \ (\text{C(2')}); 72.5 \ (\text{C(5')}); 64.9 \ (\text{CH}_2); 61.3 \ (\text{C(4')}); 26.8 \ (\text{Me}_3\text{C}); 19.1 \ (\text{Me}_3\text{C}). \ (\text{Me}_3\text{C}); 19.1 \ (\text{Me}_3\text{C})$ 

To a soln. of **11a** (0.32 g, 0.66 mmol) in MeOH (30 ml), NH<sub>4</sub>F (0.4 g) was added and the soln. heated under reflux for 3 h. The soln. was then adsorbed on silica gel and purified by CC: **13a** (0.14 g, 85%). White solid. M.p. 190°. UV (MeOH): 260.6 (14000).  $^{1}$ H-NMR ((D<sub>6</sub>)DMSO): 8.15 (2s, H-C(2), H-C(8)); 7.25 (br. s, NH<sub>2</sub>); 5.56 (t, H-C(4')); 5.30 (br. s, 1 H, =CH<sub>2</sub>); 5.15 (t, t = 2.0, 1 H, =CH<sub>2</sub>); 5.06 (t, t = 5.6, OH); 4.44 (br. s, H-C(2')); 3.73 (t = 2.0, 1 H, =C(5')); 3.31 (t = 6.7, CH<sub>2</sub>).  $^{13}$ C-NMR (CD<sub>3</sub>OD): 157.3 (C(6)); 153.7 (C(2)); 150.5 (C(4)); 148.6 (C(3')); 141.4 (C(8)); 119.9 (C(5)); 112.7 (=CH<sub>2</sub>); 83.1 (C(2')); 72.9 (C(5')); 64.2 (CH<sub>2</sub>); 58.7 (C(4')). HR-FAB-MS: 270.0959 (C<sub>11</sub>H<sub>13</sub>N<sub>3</sub>NaO<sup>2</sup><sub>7</sub>, [t + Na] $^{+}$ ; calc. 270.0966).

Compound **12a** was similarly deprotected to give **14a** (86%). White solid. M.p. 256°. UV (MeOH): 274.5 (13300).  $^1$ H-NMR (CD<sub>3</sub>OD): 8.62 (s, H-C(8)); 7.90 (s, H-C(2)); 5.85 (m, H-C(4')); 5.49 (m, CH<sub>2</sub>); 4.52 (br. s, H-C(2')); 4.29 (dd, J = 10.6, 2.7, H<sub>a</sub>-C(5')); 4.22 (dd, J = 10.6, 5.9, H<sub>b</sub>-C(5')); 4.00 (dd, J = 12.5, 2.9, 1 H, CH<sub>2</sub>); 3.90 (dd, J = 12.4, 3.3, 1 H, CH<sub>2</sub>).  $^1$ C-NMR (CD<sub>3</sub>OD): 156.6 (C(6)); 152.9 (C(2)); 150.5 (C(4)); 147.4 (C(3')); 144.3 (C(8)); 120.8 (C(5)); 114.4 (=CH<sub>2</sub>); 83.3 (C(2')); 73.1 (C(5')); 63.7 (CH<sub>2</sub>); 63.6 (C(4')). HR-FAB-MS: 270.0954 (C<sub>11</sub>H<sub>13</sub>N<sub>5</sub>NaO<sub>7</sub>\*, [M + Na]+; calc. 270.0966).

(2S,4S)-4-(7-Amino-3H-1,2,3-triazolo[4,5-d]pyrimidin-3-yl)tetrahydro-3-methylidenefuran-2-methanol (13b) and (2S,4S)-4-(7-Amino-4H-1,2,3-triazolo[4,5-d]pyrimidin-4-yl)tetrahydro-3-methylidenefuran-2-methanol (14b). Compounds 11b and 12b were synthesized as described for 13a and 13b in 50 and 23% yield, resp.

Data of **11b**: M.p.  $168^{\circ}$ . <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 8.37 (*s*, H–C(5)); 7.73 – 7.34 (*m*, 2 Ph); 7.06 (br. *s*, NH<sub>2</sub>); 5.93 (*t*, *J* = 6.3, H–C(4')); 5.25 (*s*, 1 H, =CH<sub>2</sub>); 5.04 (*s*, 1 H, =CH<sub>2</sub>); 4.74 (*t*, *J* = 5.4, H–C(2')); 4.50 (*dd*, *J* = 9.4, 6.0, H<sub>a</sub>–C(5')); 4.33 (*dd*, *J* = 9.4, 7.2, H<sub>b</sub>–C(5')); 4.10 (*dd*, *J* = 10.8, 6.6, 1 H, CH<sub>2</sub>); 3.91 (*dd*, *J* = 10.8, 5.1, 1 H, CH<sub>2</sub>); 1.07 (*s*, Me<sub>3</sub>C). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 156.5 (C(7)); 156.1 (C(5)); 149.1 (C(3a)); 146.2 (C(3')); 135.6, 133.4, 133.2, 129.6, 127.6 (2 Ph); 124.4 (C(7a)); 111.4 (=CH<sub>2</sub>); 81.5 (C(2')); 69.6 (C(5')); 65.8 (CH<sub>2</sub>); 60.0 (C(4')); 26.7 (Me<sub>3</sub>C); 19.1 ( $Me_3$ C).

Data of **12b**: M.p. 74–75°. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 8.49 (s, H–C(5)); 7.70–7.32 (m, 2 Ph); 5.81 (m, H–C(4')); 5.39 (s, 1 H, =CH<sub>2</sub>); 5.36 (s, 1 H, =CH<sub>2</sub>); 4.68 (m, H–C(2')); 4.68 (dd, J = 10.1, 4.0, H<sub>a</sub>–C(5')); 4.35 (dd, J = 10.1, 6.5, H<sub>b</sub>–C(5')); 3.97 (dd, J = 10.8, 6.3, 1 H, CH<sub>2</sub>); 3.85 (dd, J = 15.1, 5.1, 1 H, CH<sub>2</sub>). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 157.8 (C(7)); 156.6 (C(3a)); 156.4 (C(5)); 145.5 (C(3')); 135.5, 135.4, 133.2, 133.0, 129.6, 127.5 (2 Ph); 126.0 (C(7a)); 113.3 (=CH<sub>2</sub>); 81.4 (C(2')); 70.9 (C(5')); 68.9 (C(4')); 65.9 (CH<sub>2</sub>), 26.6 (Me<sub>3</sub>C); 19.0 ( $Me_3$ C).

Compounds 11b and 12b were deprotected as described for 11a to give 13b and 14b in 81 and 88% yield, resp.

Data of **13b**: White solid powder. M.p. 243°. UV (MeOH): 279.5 (12400).  $^{1}$ H-NMR ((D<sub>6</sub>)DMSO): 8.45 (br. s, 1 H, NH<sub>2</sub>); 8.29 (s, H – C(5)); 8.11 (br. s, 1 H, NH<sub>2</sub>); 5.93 (m, H – C(4')); 5.26 (s, 1 H, =CH<sub>2</sub>); 4.95 (m, 2 H, =CH<sub>2</sub>, -OH); 4.51 (br. s, H – C(2')); 4.38 (dd, J = 9.4, 5.4, H<sub>a</sub> – C(5')); 4.25 (dd, J = 9.4, 8.2, H<sub>b</sub> – C(5')); 3.74 (m, 1 H, CH<sub>2</sub>); 3.59 (m, 1 H, CH<sub>2</sub>).  $^{13}$ C-NMR ((D<sub>6</sub>)DMSO): 156.7 (C(7)); 156.2 (C(5)); 148.6 (C(3a)); 147.1 (C(3')); 123.9 (C(7a)); 110.4 (=CH<sub>2</sub>); 81.5 (C(2')); 68.9 (C(5')); 63.4 (CH<sub>2</sub>); 59.6 (C(4')). HR-FAB-MS: 271.0902 (C<sub>10</sub>H<sub>12</sub>N<sub>8</sub>NaO<sub>7</sub><sup>+</sup>, [M + Na]<sup>+</sup>; calc. 271.0919).

Data of **14b**: White powder. M.p. 201°. UV (MeOH): 295.3 (10900). ¹H-NMR (CD<sub>3</sub>OD): 8.28 (s, H-C(5)); 5.93 (m, H-C(4')); 5.41 (d, J = 4.5, =CH<sub>2</sub>); 4.65 (dd, J = 10.3, 3.3, H<sub>a</sub>-C(5')); 4.58 (m, H-C(2')); 4.33 (dd, J = 10.3, 6.3, H<sub>b</sub>-C(5')); 3.82 (dd, J = 11.7, 6.7, 1 H, CH<sub>2</sub>); 3.73 (dd, J = 11.7, 4.1, CH<sub>2</sub>). ¹³C-NMR (CD<sub>3</sub>OD): 158.7 (C(7)); 158.6 (C(3a)); 157.7 (C(5)); 147.6 (C(3')); 127.1 (C(7a)); 113.6 (=CH<sub>2</sub>); 83.2 (C(2')); 72.1 (C(5')); 70.6 (C(4')); 65.1 (CH<sub>2</sub>). HR-FAB-MS: 271.0919 (C<sub>10</sub>H<sub>12</sub>N<sub>6</sub>NaO<sub>2</sub>+, [M + Na]+, calc. 271.0919).

3-Benzoyl-1-{(3S,5S)-5-{{[(tert-butyl)diphenylsilyl]oxy}methyl}tetrahydro-4-methylidenefuran-3-yl}-5-methyl pyrimidine-2,4(1H,3H)-dione (15a). To a soln. of Ph<sub>3</sub>P (0.3 g, 0.92 mmol) in anh. THF (10 ml), DEAD (0.17 ml, 1.1 mmol) was added. The mixture was stirred at 0° for 30 min and then cooled to  $-45^{\circ}$ . To this soln., N²-benzoylthymine (0.21 g, 0.92 mmol) and 10b (0.17 g, 0.46 mmol) in THF (10 ml) were added within 1 h. The mixture was stirred overnight at  $-45^{\circ}$ . The soln. was directly adsorbed on silica gel and purified by Cc: 15a (0.08 g, 30%). Low-melting solid. ¹H-NMR (CD<sub>3</sub>OD): 7.92 – 7.25 (m, H – C(6), 3 Ph); 5.66 (br. s, H – C(3')); 5.34 (d, J = 13.6, = CH<sub>2</sub>); 4.47 (br. s, H – C(5')); 4.14 (dd, J = 10.2, 6.8, 1 H); 4.04 (dd, J = 11.0, 3.7, 2 H); 3.94 (dd, J = 11.3, 4.5, 1 H); 1.64 (s, Me); 1.10 (s, Me<sub>3</sub>C). ¹³C-NMR (CDCl<sub>3</sub>): 168.9 (C=O); 162.5 (C(4)); 150.3 (C(2)); 146.3 (C(4')); 136.8 (C(6)); 112.1 (= CH<sub>2</sub>); 111.6 (C(5)); 81.1 (C(5')); 71.1 (C(2')); 65.4 (CH<sub>2</sub>); 57.3 (C(3')); 26.9 (Me<sub>3</sub>C); 19.3 ( $Me_3$ C); 12.2 (Me). HR-FAB-MS: 603.2305 (C<sub>34</sub>H<sub>36</sub>N<sub>2</sub>NaO<sub>5</sub>Si<sup>+</sup>, [M + Na]<sup>+</sup>; calc. 603.2291).

5-Methyl-1-[(3S,5S)-tetrahydro-5-(hydroxymethyl)-4-methylidenefuran-3-yl]pyrimidine-2,4(1H,3H)-dione (16a). To a soln. of 15a (0.07 g, 0.13 mmol) in MeOH (10 ml), NH<sub>4</sub>F (0.065 g, 2 mmol) was added, and the soln. was

heated under reflux for 3 h. The soln. was cooled to r.t., adsorbed on silica gel, and purified by CC to give the N(3)-benzoylated derivative, which was dissolved in MeOH (5 ml) and treated with NaOMe (0.03 g). The mixture was stirred at r.t. for 8 h, neutralized with 10% AcOH/H<sub>2</sub>O, and evaporated. The residue was taken up in MeOH, adsorbed on silica gel, and purified by CC: **16a** (0.018 g, 62% for two steps). Hygroscopic foam. UV (MeOH): 270.9 (9400). <sup>1</sup>H-NMR ((D<sub>6</sub>)acetone): 10.14 (br. s, NH); 7.62 (s, H-C(6)); 5.56 (m, H-C(3')); 5.36 (t, t = 2.1, 1 H, = CH<sub>2</sub>); 5.30 (t, t = 2.3, 1 H, = CH<sub>2</sub>); 4.40 (br. s, H-C(5')); 4.18 – 3.84 (t = 3.84 (t = 4.70 (C(2)); H-C(3), CH<sub>2</sub>); 1.76 (t = 4.80 (C(5')); 72.2 (C(2')); 638 (CH<sub>2</sub>); 58.9 (C(3')); 12.3 (Me). HR-FAB-MS: 261.0855 (C<sub>11</sub>H<sub>14</sub>N<sub>2</sub>NaO<sub>4</sub>+, [t t + Na]+; calc. 261.0851).

1-[(3S,5S)-Tetrahydro-5-(hydroxymethyl)-4-methylidenefuran-3-yl]pyrimidine-2,4(1H,3H)-dione (16b). To a soln. of Ph<sub>3</sub>P (0.84 g, 3.18 mmol) in anh. THF, DEAD (0.49 ml, 3.12 mmol) was added. The mixture was then stirred at 0° for 30 min and cooled to −70°. A suspension of 3-benzoyluracil (0.54 g, 3.12 mmol) and 10b (0.46 g, 1.25 mmol) was added to the soln. within 30 min. The mixture was stirred overnight at −40° and then quenched with  $H_2O$  (1 ml) and evaporated. The gummy residue was purified by CC. The product was dissolved in MeOH (40 ml), NH<sub>4</sub>F (0.5 g) added, and the resulting soln. heated under reflux for 3 h. The soln. was directly adsorbed on silica gel and purified by CC: 16b (0.16 g, 57%). White solid. M.p. 178°. UV (MeOH): 266.6 (11800).  $^{1}$ H-NMR (CD<sub>3</sub>OD): 7.80 (d, J = 8.1, H−C(6)); 5.64 (d, J = 8.1, H−C(5)); 5.57 (m, H−C(3')); 5.35 (m, =CH<sub>2</sub>); 4.82 (br. s, H−C(5')); 4.07 (dd, J = 10.3, 6.5, H<sub>a</sub>−C(2')); 4.00 (dd, J = 10.2, 3.6, H<sub>b</sub>−C(2')); 3.90 (dd, J = 12.3, 2.9, 1H, CH<sub>2</sub>); 3.80 (dd, J = 12.3, 4.2, 1 H, CH<sub>2</sub>).  $^{13}$ C-NMR (CD<sub>3</sub>OD): 166.2 (C(4)); 153.2 (C(2)); 148.5 (C(4')); 144.6 (H−C(6)); 112.5 (=CH<sub>2</sub>); 102.7 (C(5)); 82.9 (C(5')); 72.3 (C(2')); 63.9 (CH<sub>2</sub>); 59.3 (C(3')). HR-FAB-MS: 247.0703 (C<sub>10</sub>H<sub>1</sub>)×N,NaO<sub>4</sub>+, [M + Na]+; calc. 247.0694).

4-Amino-1-[(3S,5S)-tetrahydro-5-(hydroxymethyl)-4-methylidenefuran-3-yl]pyrimidin-2(IH)-one (17). Ac<sub>2</sub>O (0.25 ml, 2.65 mmol) was added to a soln. of 16b (0.12 g, 0.53 mmol) in pyridine (10 ml) at r.t. After the addition, the mixture was stirred at r.t. overnight and then quenched with  $H_2O$  (1 ml). The solvent was evaporated and the conc. soln. co-evaporated with toluene. The residue was purified by CC to give the acetylated derivative (0.13 g, 92%) as pale yellow foamy solid. Et<sub>3</sub>N (0.1 ml) was added to a soln. of the acetylated derivative (0.09 g, 0.33 mmol) in MeCN (10 ml) containing  $2,4,6^{-1}Pr_3C_6H_2SO_2Cl$  (0.23 g, 0.75 mmol) and 4-(dimethylamino)pyridine (0.092 g, 0.75 mmol) at 0°. The mixture was stirred at r.t. for 4 h. NH<sub>4</sub>OH soln. (28%, 6 ml) was added, and the soln. was stirred at r.t. for 24 h. After evaporation, the residue was purified by CC: 17 (0.035 g, 47% in two steps). Hygroscopic solid. UV (MeOH): 276.4 (8650). <sup>1</sup>H-NMR (CD<sub>3</sub>OD): 7.80 (d, J = 7.4, H -C(6)); 5.83 (d, J = 7.5, H -C(5)); 5.63 (m, H -C(3')); 5.35 (t, J = 2.1, 1 H, -CH<sub>2</sub>); 5.30 (t, J = 2.2, 1 H, -CH<sub>2</sub>); 4.40 (br. s, H -C(5')); 4.08 ( $dd, J = 10.1, 6.5, H_a$  -C(2')); 3.96 ( $dd, J = 10.1, 3.6, H_b$  -C(2')); 3.90 (dd, J = 12.3, 3.0, 1 H, CH<sub>2</sub>); 3.81 (dd, J = 12.3, 4.2, 1 H, CH<sub>2</sub>). <sup>13</sup>C-NMR (CD<sub>3</sub>OD): 167.4 (C(2)); 159.2 (C(4)); 149.0 (C(4')); 144.8 (C(6)); 112.2 (-CH<sub>2</sub>); 96.1 (C(5)); 83.1 (C(5')); 72.9 (C(2')); 63.9 (CH<sub>2</sub>); 60.3 (C(3')). HR-FAB-MS: 224.1040 (C<sub>10</sub>H<sub>14</sub>N<sub>3</sub>O<sub>3</sub>+; [M+H]+; calc. 224.1035).

2-{(3\$,5\$)-5-{[[(tert-Buyl)diphenylsilyl]oxy]methyl]tetrahydro-4-methylidenefuran-3-yl]-1H-isoindol-1,3(2H)-dione (19). To a soln. of 10b (1.1 g, 3 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (30 ml), Et<sub>3</sub>N (1.7 ml, 12 mmol) and MsCl (0.7 ml, 9 mmol) were added at 0°. After the addition, the mixture was stirred at 0° for 3 h. H<sub>2</sub>O (75 ml) was added and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 30 ml). The combined CH<sub>2</sub>Cl<sub>2</sub> part was washed with H<sub>2</sub>O, dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated. The oily residue was purified by CC: 18 (1.11 g, 83%). Pale yellow low-melting solid. ¹H-NMR (CDCl<sub>3</sub>): 7.70 (m, 4 H, Ph); 7.42 (m, 6 H, Ph); 5.64 (s, 1 H, =CH<sub>2</sub>); 5.51 (m, H-C(3)); 5.33 (s, 1 H, =CH<sub>2</sub>); 4.65 (br. s, H-C(5)); 4.18 (dd, J = 10.6, 4.8, H<sub>a</sub>-C(2)); 4.06 (dd, J = 10.6, 2.8, H<sub>b</sub>-C(2)); 3.82 (dd, J = 11.0, 3.7, 1 H, CH<sub>2</sub>); 3.71 (dd, J = 11.0, 4.6, 1 H, CH<sub>2</sub>); 3.04 (s, Me); 1.06 (s, Me<sub>3</sub>C). ¹³C-NMR (CDCl<sub>3</sub>): 145.4 (C(4)); 114.3 (=CH<sub>2</sub>); 80.9, 80.2 (C(3), C(5)); 72.1 (C(2)); 66.3 (CH<sub>2</sub>); 39.2 (Me); 26.9 (Me<sub>3</sub>C); 19.3 ( $Me_3$ C).

To a soln. of **18** (0.3 g, 0.67 mmol) in DMF (30 ml), potassium phthalimide (0.74 g, 4 mmol) was added, and the mixture was heated at 110° overnight. The soln. was cooled and filtered, and the filtrate was evaporated. The residue was taken up in AcOEt (80 ml) and washed with  $H_2O$  (2 × 50 ml). The AcOEt part was dried (Na<sub>2</sub>SO<sub>4</sub>), evaporated, and purified by CC: **19** (0.17 g, 51%). Low-melting solid. <sup>1</sup>H-NMR (CDCl<sub>3</sub>): 7.84 – 7.30 (m, 14 arom. H); 5.34 (m, H – C(3')); 5.03 (t, J = 2.2, 1 H, = CH<sub>2</sub>); 4.87 (t, J = 2.4, 1 H, = CH<sub>2</sub>); 4.70 (m, H – C(5')); 4.15 (dd, J = 11, 7.8, 1 H, CH<sub>2</sub>); 4.04 (d, J = 8.7, 2 H – C(2')); 3.86 (dd, J = 11.1, 4.6, 1 H, CH<sub>2</sub>); 1.11 (s, Me<sub>3</sub>C). <sup>13</sup>C-NMR (CDCl<sub>3</sub>): 167.1 (C=O); 145.8 (C(4')); 135.7, 135.6, 134.1, 133.8, 133.5, 131.7, 129.5, 127.6, 123.3 (arom. C); 107.0 (=CH<sub>2</sub>); 81.8 (C(5')); 66.0, 65.9 (C(2'), CH<sub>2</sub>); 51.9 (C(3')); 26.8 (Me<sub>3</sub>C); 19.2 ( $Me_3$ C). HR-FAB-MS: 520.1930 (C<sub>30</sub>H<sub>31</sub>NNaO<sub>4</sub>Si<sup>+</sup>, [M + Na]<sup>+</sup>; calc. 520.1920).

(3S,5S)-5-[[[(tert-Butyl)diphenylsilyl]oxy]methyl]tetrahydro-4-methylidenefuran-3-amine (20). To a soln. of 19 (0.16 g, 0.32 mmol) in EtOH (20 ml), hydrazine hydrate (0.12 g) was added, and the mixture was heated under reflux for 3 h and then allowed to cool to r.t. The white precipitate was filtered and washed with EtOH.

The combined EtOH part was evaporated and the residue purified by CC: **20** (0.09 g, 77%).  $^{1}$ H-NMR (CDCl<sub>3</sub>): 7.71 – 7.25 (m, 2 Ph); 5.29 (br. s, 1 H, =CH<sub>2</sub>); 4.48 (m, H-C(5)); 3.99 – 3.64 (m, 2 H-C(2), H-C(3), CH<sub>2</sub>); 1.06 (s, Me<sub>3</sub>C).  $^{13}$ C-NMR (CDCl<sub>3</sub>): 153.3 (C(4)); 135.6, 133.3, 133.2, 129.6, 127.6 (Ph); 106.2 (=CH<sub>2</sub>); 81.0 (C(5)); 74.4 (C(2)); 66.4 (CH<sub>2</sub>); 55.5 (C(3)); 26.8 (Me<sub>3</sub>C); 19.1 ( $Me_3$ C). HR-FAB-MS: 390.1870 (C<sub>22</sub>H<sub>29</sub>NNaO<sub>2</sub>Si<sup>+</sup>, [M + Na]<sup>+</sup>; calc. 390.1865).

5-Methyl-1-[(3S,5S)-tetrahydro-5-(hydroxymethyl)-4-methylidenefuran-3-yl]pyrimidine-2,4(1H,3H)-dione (16a). A soln. of 3-methoxy-2-methylprop-2-enoyl isocyanate (prepared by heating the corresponding acid chloride (0.11 g, 0.8 mmol) and silver cyanate (0.22 g) in toluene for 0.5 h) was added dropwise to a soln. of 20 (0.09 g, 0.25 mmol) in DMF (10 ml) and  $\rm Et_2O$  (5 ml) at  $-10^\circ$ . After the addition, the mixture was allowed to warm to r.t. and stirred for 16 h. The volatile materials were evaporated. EtOH was added to the residue with stirring and then evaporated. The gummy residue was purified by CC to give the acryloylurea derivative 21. This compound was dissolved in dioxane (5 ml) and heated at  $100^\circ$  for 4 h with  $2 \rm NH_2SO_4$  (1.0 ml). The mixture was allowed to cool to r.t. and then neutralized with  $2 \rm NH_2SO_4$  and evaporated. The residue was purified by CC: 16a (0.04 g, 50% in two steps), identical to the compound prepared by direct coupling and deprotection (see above).

Single-Crystal X-Ray Structure Determination of Compounds 13a and 14a. A colorless prismatic crystal  $(0.21 \times 0.24 \times 0.36 \text{ mm} \text{ and } 0.25 \times 0.25 \times 0.31 \text{ mm}, \text{ resp.})$  was isolated from the sample and mounted with grease on the tip of a glass capillary epoxied to a brass pin and placed on the diffractometer with the long crystal dimension (a-axis and AB dimension, resp.) approximately parallel to the diffractometer  $\phi$ -axis. Data were collected on an Enraf-Nonius-CAD4 diffractometer (MoKa radiation, graphite monochromator) at 200 K (cold  $N_2$  gas cooling) using  $\theta$ -2 $\theta$  scans. Intensity standards were measured at 2 h intervals. Net intensities were obtained by profile analysis of the 4469 and 5428 data, resp. Lorentz and polarization corrections were applied. No change in the intensity standards was detected. Absorption was minimal and no correction was applied. Equivalent data were averaged yielding 2018 unique data (R-int = 0.023, 1926 > 4  $\sigma$  (F)) in the case of 13a and 3817 unique data (R-int = 0.072, 3042 >  $4\sigma(F)$ ) in the case of **14a**. Based on preliminary examination of the data, the space group P(1)(1)(1)(1) was assigned to **13a** and P(1) to **14a**. The computer programs from the MoLEN package were used for data reduction. The preliminary model of the structure was obtained with XS, a directmethods program. Least-squares refining of the model vs. the data was performed with the XL computer program. Illustrations were made with the XP program, and tables were made with the XCIF program. All are in the SHELXTL v5.0 package. Thermal ellipsoids are drawn at the 35% level unless otherwise noted. All non-Hatoms were refined with anisotropic thermal parameters. All H-atoms were included with a riding model based on default values. In the case of 14a, there are two independent molecules in the asymmetric unit. Each forms a dimer via two H-bonds with a symmetry-equivalent molecule of the other (i.e., molecule A dimerizes with molecule B, and vice versa). The two molecules differ in conformation mainly at the sugar CH<sub>2</sub>OH group.

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